## Highly Enantioselective Total Synthesis of (+)-Isonitramine

## Yohan Park,† Young Ju Lee,‡ Suckchang Hong,‡ Myungmo Lee,‡ and Hyeung-geun Park\*

College of Pharmacy, Inje University, 607 Obang-dong, Gimhae, Gyeongnam 621-749, Korea, and Research Institute of Pharmaceutical Sciences and College of Pharmacy, Seoul National Univerisity, Seoul 151-742, Korea

hgpk@snu.ac.kr

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A new efficient enantioselective synthetic method of  $(+)$ -isonitramine is reported.  $(+)$ -Isonitramine was obtained in 12 steps (98% ee and 43%) overall yield) from δ-valerolactam via enantioselective phase-transfer catalytic alkylation, Dieckman condensation, and diastereoselective reduction as key steps.

There are a number of biologically important natural  $p_{\text{roducts}}^1$  containing optically active 2-azaspirocycle structures, such as polyzonimine,<sup>1a</sup> nitropolyzonamine,<sup>1a</sup> horsfiline,<sup>1b</sup> and spirotryprostatin B.<sup>1c</sup> Among these compounds, *Nitraria* alkaloids,  $^{2}(+)$ -nitramine (1),

† Inje University.

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 $(+)$ -isonitramine (2), and  $(-)$ -sibirine (3), have chiral quaternary carbon centers on the 2-azaspiro[5,5]undecane-7-ol skeletons. The structural similarity with the neurotoxic (-)-histrionicotoxin has faciliated the development of efficient synthetic routes of Nitraria alkaloids (Figure 1). Recently, it was reported that the extracts of Nitraria plants have antiproliferative effects on cancer cell lines through the apoptosic pathway.<sup>3</sup>



Although numerous enantioselective synthetic methods for *Nitraria* alkaloids have been reported thus far,<sup>4</sup> most of these approaches employed chiral auxiliaries, $4a,d,j$  chiral substrates,  $4b, g$  or enzymatic resolution,  $4e, f, h, l$  which cannot be readily applied to large scale production. In addition,

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<sup>‡</sup> Seoul National University.

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only one reported catalytic method uses an organopalladium catalyst for the establishment of quaternary carbon center  $(6R)$ ; however, the enantioselectivity is only  $86\%$ ee.<sup>4i</sup> As one of our research programs for the development of new therapeutics for the treatment of cancer, we need to develop very efficient and highly enantioselective synthetic methods for Nitraria alkaloids. Herein, we present a novel approach to synthesize  $(+)$ -isonitramine (2), a representative Nitraria alkaloid, via asymmetric phase-transfer catalytic (PTC) alkylation, which is regarded as one of the most efficient synthetic processes for large scale production from the view points of economical and environmental aspects.<sup>5</sup>



Very recently, we reported on the highly enantioselective PTC alkylations of  $\alpha$ -tert-butoxycarbonylactams (Scheme 1).<sup>6</sup> The alkylation of  $N(1)$ -methyl-2-oxo-pyrrolidine-3-carboxylic acid tert-butyl ester or N(1)-benzhydryl-2-oxo-piperidine-3-carboxylic acid tert-butyl ester under PTC conditions in the presence of chiral quaternary ammonium salts successfully afforded the corresponding  $\alpha$ -alkylated lactams in high chemical (up to 99%) and optical yields (up to 98% ee), which is potencially applicable to the synthesis of 2-azaspirocycles by further cyclization. Given this, we attempted to apply our novel method to the synthesis of a representative *Nitraria* alkaloid,  $(+)$ -isonitramine  $(2)$ .

As shown in a retrosynthetic analysis (Scheme 2), the  $C(7R)$  chirality can, in principle, be induced by diastereoselective reduction of 6, which can be obtained by Dieckmann condensation<sup>7</sup> of 5. Optically active  $(R)$ -5 can be derived from the enantioselective phase-transfer catalytic alkylation of N(1)-benzhydryl-2-oxo-piperidine-3-carboxylic acid tertbutyl ester (4).

First, the substrate (4) for PTC alkylation was prepared from  $\delta$ -valerolactam (7) in two steps (Scheme 3). N-Diphenylmethylation of 4 under PTC reaction conditions in the presence of TBAB as a catalyst, followed by tertbutylcarboxylation using Boc<sub>2</sub>O under LiHMDS basic conditions, provided lactam 4 in good yield.

Since the phase-transfer catalytic alkylation of 4 with unactivated alkyl halides usually produced poor chemical Scheme 2. Retrosynthetic Analysis



Scheme 3. Enantioselective PTC Alkylation



yields, we changed the synthetic route of 5 through allylation, followed by chain elongation. However, the allylation still resulted in unsatisfactory enantioselectivity (87% ee) as reported previously; thus, we attempted to improve the enantioselectivity. We previously improved the enantioselectivity in the monoallylation of the N,N-diphenylmalonamide ester system during the total synthesis of  $(-)$ -paroxetine by switching the "allyl bromide" with "2bromoallyl bromide".8 Therefore, we used bromoallyl bromide as an alternative electrophile. The phase-transfer catalytic allylation of 4 was carried out with 2-bromoallyl

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**Scheme 4.** Synthesis of  $(+)$ -Isonitramine<sup>a,b</sup>



<sup>a</sup>The enantiopurity was determined by HPLC analysis of 11, 12, and 16 using a chiral column (Chiralpak AD-H) with hexanes/2-propanol as the eluent. <sup>b</sup>Absolute configuration was assigned by comparison of the specific optical rotation value of the  $(+)$ -isonitramine (2) with the literature value.<sup>4f</sup>

bromide under the previously reported optimal reaction conditions<sup>6</sup>  $[(S, S)$ -3,4,5-Trifluorophenyl-NAS bromide  $(9, 5 \text{ mol } \%)$ , and solid KOH  $(5.0 \text{ equiv})$  in toluene at  $-40$  °C] to afford 11. Surprisingly, the enantioselectivity was dramatically increased (87% ee to 98% ee), as well as the chemical yield (87% to 95%). These findings indicate that the bulkier electrophile was better able to selectively approach the re-face of the enolate of 4 when complexed with quaternary ammonium catalyst 10, resulting in  $C(6R)$ chirality with higher enantioselectivity. The debromination of 11 using tributyltin hydride in the presence of a catalytic amount of AIBN in toluene gave  $12$  (95%) (Scheme 4).<sup>9</sup> For the elongation of the carbon chain, ethyl ethoxythiocarbonylsulfanylacetate was successfully added to the allylic moiety of 12 under radical conditions to provide 13  $(88\%)$ .<sup>10,11</sup>The xanthate was then removed using tributyltin hydride and AIBN, which afforded diester 5 (96%). Since the direct Dieckmann condensation of 5 for the construction of the spirocyclic skeleton failed, the tert-butyl ester of 5 was transformed to afford the corresponding methyl ester 14 by the removal of tert-butyl ester with TFA in methylene chloride, followed by methylation with an excess of diazomethane (98%). Dieckmann condensation of 14 was then successfully accomplished under LiHMDS base conditions in THF at  $0^{\circ}$ C, resulting in the production of spirocyclic  $\beta$ -ketoester 15 (98%). The hydrolysis of 15 with 50% aqueous KOH in MeOH, followed by decarboxylation, afforded ketone 6 (93%).

Next, the  $C(7R)$  chirality of 2 was introduced by the diastereoselective reduction of 6. The reduction of 6 using

DIBAL in THF at  $-78$  °C afforded 16 as a single diastereomer of the desired 7R-form in 93% chemical yield. Notably, the diastereoselectivity was much higher than that in the reduction of the free amide analogue of 6 under the same reaction conditions, which was reported previously  $(87\%$  de).<sup>4f</sup> The bulky benzhydryl group seems to play important roles in not only the high enantioselectivity of  $C(6R)$  in the PTC alkylation of 4 but also the high enantioselectivity of  $C(7R)$  in the diastereoselective reduction of 6. The reduction of the lactam 16 using excess LiAlH<sub>4</sub> in THF readily afforded piperidine  $17 (90\%)$ . Finally,  $(+)$ -isonitramine {2,  $[\alpha]_D^{23} = +6.22$  (c 1.16, CHCl<sub>3</sub>);  $[\alpha]_D^{23} = +5.8$  $(c \ 1.4, CHCl<sub>3</sub>)<sup>4f</sup>$  could be obtained by the catalytic hydrogenation of 17 with  $Pd(OH)_2$  in methanol (87%).

In conclusion, a new efficient synthetic approach toward  $(+)$ -isonitramine (2) was developed.  $(+)$ -Isonitramine (2) was synthesized in 12 steps (43% overall yield, 98% ee) by enantioselective phase-transfer catalytic alkylation, Dieckmann condensation, and diastereoselective reduction from  $\delta$ -valerolactam (7). Both the high enantioselectivity and chemical yield make this approach a practical route for the large scale synthesis of 2-azaspirocycles. Structure-activity relationship studies of  $(+)$ -isonitramine derivatives are now in progress.

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Supporting Information Available. Spectroscopic characterizations of compounds  $2, 4-6, 8,$  and  $10-17$ . This material is available free of charge via the Internet at http://pubs.acs.org.

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